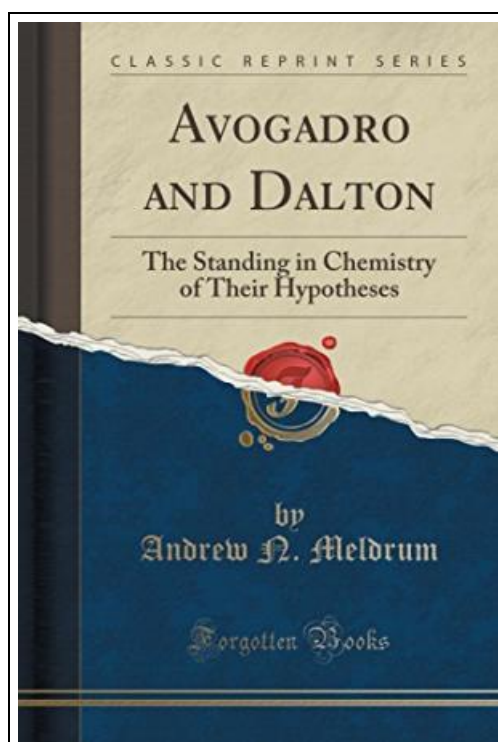


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Forgotten Books, United States, 2015. Paperback. Book Condition: New. 229 x 152 mm. Language: English . Brand New Book ***** Print on Demand *****.Excerpt from Avogadro and Dalton: The Standing in Chemistry of Their Hypotheses The important question whether, in discussing the constitution of matter, the atom or the molecule should be first considered, is one regarding which there has been considerable difference of opinion among chemists. Formerly, few would have disputed the claim of the atom to prior consideration. Not only did Dalton's atomic hypothesis precede Avogadro's molecular hypothesis in order of time, but this chronological order seemed also to be the natural order, involving, as it did, only the usual transition from the simpler to the more complex conception. But the history of the science shows that, enormous as were the services which Dalton's atomic hypothesis rendered to chemistry, the chief object of that hypothesis - the determination of a set of consistent atomic weights - remained for a long time merely a pious hope. Avogadro's rule supplied a means of determining the molecular weights of substances, and, from these molecular weights, of ascertaining which of several possible atomic weights of a contained element was the correct one. Until this step had been taken, the marvellous developments of theoretical chemistry which have characterised the last fifty years would have been impossible. In the study of the constitution of matter, therefore, we are compelled, by the very nature of the particular problems of scientific measurement and calculation involved, to reverse what is apparently the natural order of things and to proceed from the more complex to the simpler - from highly complex objects of sense to successively simpler and simpler conceptual structures underlying these. Thus the order of quantitative determination is: - (1) relative weights...

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