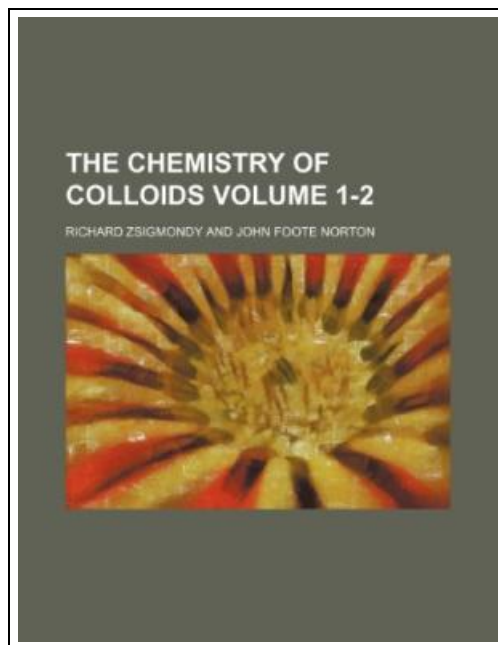


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1917 Excerpt: .the nature of sulfuric acid. Precipitation with Electrolytes.-- As already pointed out silicic acid is not immediately precipitated by all electrolytes. Hydrochloric acid, chlorides of the alkali, and alkaline earth metals give no precipitate but may cause gelatinization after standing for hours, days, or perhaps weeks. It is precipitated immediately by barium hydroxide, - E. Jordis: Zeit. f. anorg. Chemie, 36, 16-22 (1903). t R. Zsigmondy und R. Heyer: Zeit. f. anorg. Chemie, 68, 169-187 (1910). t A. Sabanejeff: Journ. d. russ. phys.-chem. Ges., 21, 515-525 (1889); Ber., 23, R., 87 (1890). G. Bruni e N. Pappada: Atti della R. Accad. dei Lincei Roma (5), 9, 354-358 (1900); Gazzetta chimica ital., 31, 1, 244-252 (1901). concentrated solutions of ammonium sulfate, dilute solutions of egg albumin, glue, and many basic dyestuffs such as Methylene Blue. Regarding the effect of the salts of the alkali metals on the precipitation, Pappada found that the anions played little part but that the influence of the cations increased with the molecular weight, caesium having the greatest and lithium the least. Protective Influence of Silicic Acid.--Silicic acid hydrosol does not act as a protective colloid for colloidal gold. It does not hinder the change of color by sodium chloride, neither does it prevent the formation of a turbidity nor a precipitate in silver chloride solutions, unless the amount of chloride ion is extremely small. In the latter case the reaction is obscure, and is not protective action in the ordinary sense. This is...



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